EXCIPIENT FORMULATION AND PROCESSING GUIDE FOR ORAL LIQUID AND TOPICAL DOSAGE FORMS
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Introduction
Lubrizol Life Science Health (LLS Health) specializes in pharmaceutical partnership and helping clients from idea to execution by providing best-in-class excipients for pharmaceutical and over-the-counter products. When you partner with LLS Health, you benefit from working with us at every stage in the development process, with the ultimate goal of creating solutions that improve patient lives.

LLS Health is one of the world’s largest manufacturer of pharmaceutical grade carbomers, polycarbophil, and thermoplastic polyurethanes and has been manufacturing pharmaceutical excipients for more than 35 years.

Our Carbopol® and Pemulen™ polymers and Noveon® polycarbophil are excipients of acrylic acid, chemically crosslinked with polyalkenyl alcohols or divinyl glycol. These polymers have been successfully formulated into a variety of commercial products, including topicals, oral suspensions and solutions, bioadhesive applications, and oral care applications.

Lubrizol works with all relevant regulatory bodies to establish and maintain the global pharmacopeial status of its pharmaceutical ingredients. In addition, Lubrizol supports its pharmaceutical ingredients with Drug Master Files (DMFs) in the United States and Europe.

Key Benefits of Lubrizol Pharmaceutical Polymers

Versatile and Efficient

- At low usage levels, Lubrizol polymers provide highly effective rheology modification for aqueous, anhydrous and hydroalcoholic systems and improved stability for suspensions and emulsions:

<table>
<thead>
<tr>
<th>Application</th>
<th>Typical Usage Level (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Topical Gels, Lotions, Creams</td>
<td>0.5-3.0</td>
</tr>
<tr>
<td>Topical Emulsions</td>
<td>0.2-0.4</td>
</tr>
<tr>
<td>Oral Suspensions/Solutions</td>
<td>0.2-1.0</td>
</tr>
</tbody>
</table>

- Excellent bioadhesive characteristics: maximize contact with biological tissues; may result in lower active usage concentrations due to enhanced active pharmaceutical ingredient (API) bioavailability.

- Compatible with most acidic, basic and neutral drugs, and well-suited for applications across a broad pH range (4.0-10.0).
Multifunctional and Stable

- Effective as primary emulsifiers or emulsion stabilizers for topical formulations.

- Thickening, suspending and emulsifying properties result in topical formulations that exhibit excellent aesthetics and skin feel.

- Various polymer grades offer diverse texture, rheology profile and different sensory attributes (ranging from light gels with fresh watery feel and quick break to heavier formulations with creamy aesthetics with long play time).

- Polymers provide a structural network for the permanent, stable suspension of insoluble active or inactive ingredients.

- Ideal for pouring, pumping, spreading and spraying applications due to the shear thinning effects of the polymer.

Safe and Reliable

- Carbopol and Pemulen polymers and Noveon polycarbophil, as a class, have received extensive review and toxicological evaluation. The safety of these polymers is supported by a history of more than 50 years of successful use in commercial products.

- Polymers in the dry powder state are chemically and thermally stable under typical storage and processing conditions.

- Temperature variations have a minimal impact on the viscosity of Carbopol polymer dispersions.

- Dispersions of Carbopol polymers are freeze-thaw stable and may be sterilized.

- Formulations such as those for “nonspill” pediatric products benefit from higher viscosity and yield values, even at low polymer concentrations.

- Taste-masking characteristics can reduce or eliminate the negative effects of bitter-tasting drugs.

*Additional information concerning the toxicological testing of Lubrizol polymers is included in the Toxicology Studies and Regulatory Information Bulletin at www.lubrizol.com/Health.
Recommended Polymers for Liquid and Semisolid Dosage Forms

A list of Lubrizol polymers that are recommended for use in liquid and semisolid dosage forms is included in Table 2.

Table 2. Lubrizol Polymers for Liquid and Semisolid Applications

| Product Trade Name | Polymerization Solvent | Application Type | Viscosity Specification Ranges
<table>
<thead>
<tr>
<th></th>
<th></th>
<th></th>
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</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Lotions</td>
<td>Creams</td>
</tr>
<tr>
<td>Carbopol® Polymers</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>971P NF</td>
<td>Ethyl Acetate</td>
<td>*</td>
<td>—</td>
</tr>
<tr>
<td>974P NF</td>
<td>Ethyl Acetate</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>980 NF</td>
<td>Cosolvent3</td>
<td>—</td>
<td>*</td>
</tr>
<tr>
<td>981 NF</td>
<td>Cosolvent3</td>
<td>*</td>
<td>—</td>
</tr>
<tr>
<td>5984 EP</td>
<td>Cosolvent3</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>ETD 2020 NF</td>
<td>Cosolvent3</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>Ultrez 10 NF</td>
<td>Cosolvent3</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>934 NF2</td>
<td>Benzene</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>934P NF2</td>
<td>Benzene</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>940 NF2</td>
<td>Benzene</td>
<td>—</td>
<td>*</td>
</tr>
<tr>
<td>941 NF2</td>
<td>Benzene</td>
<td>*</td>
<td>—</td>
</tr>
<tr>
<td>1342 NF3</td>
<td>Benzene</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>Pemulen™ Polymers</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>TR-1 NF</td>
<td>Cosolvent3</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>TR-2 NF</td>
<td>Cosolvent3</td>
<td>*</td>
<td>*</td>
</tr>
<tr>
<td>Noveon® Polycarbophil USP</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>AA-1 USP</td>
<td>Ethyl Acetate</td>
<td>—</td>
<td>*</td>
</tr>
</tbody>
</table>

1 Dispersion neutralized in water. Method information is included in the specification sheet for each product at https://www.lubrizol.com/Health/Pharmaceuticals/Resource-Hub.

2 Due to regulatory restrictions on the use of benzene in pharmaceutical formulations, Lubrizol recommends the use of carbomers polymerized in toxically preferred solvents for all new drug development projects. A list of substitute products for benzene-grade Carbopol polymers is included below. Ingredient substitutions in existing pharmaceutical formulations should be verified to confirm compliance with key performance properties and regulatory requirements.

3 Cosolvent = mixture of Ethylacetate (class III solvent) and Cyclohexane (class II solvent).
Polymer Functionality

Thickening Properties

Rheology Control Swelling and Thickening Characteristics

At low usage levels (0.2 to 3.0 wt%), Lubrizol polymers perform as highly efficient thickening agents. Polymer concentration, pH and the degree of hydrogen bonding can be modified to customize the rheological properties of the finished product for a wide range of applications.

Lubrizol’s use of different crosslinking materials and levels results in polymer types that exhibit different rheological characteristics. For example, polymers that are highly crosslinked as Carbopol 980 NF, 5984 EP, 974P NF, 934P NF, 940 NF and 934 NF exhibit very high viscosities. Alternatively, Carbopol 981 NF, 971P NF and 941 NF are lower in viscosity due to their lightly crosslinked structure.

Impact of Polymer Concentration and Type

Higher viscosities can be achieved at specific pH values by increasing the polymer concentration. This is particularly effective when targeting the rheological properties of formulations with pH values of below 5 or greater than 9. Figure 1 provides an example of the effect of polymer concentration on viscosity at constant pH.

Figure 1. Effect of Polymer Type and Concentration on the Viscosity at pH 6 - Topical Products

Figure 2. Effect of pH and Concentration on the Viscosity of Carbopol 980 NF Polymer Dispersion

Effect of pH on Properties

In aqueous systems, the primary mechanism for rheology modification is through pH adjustment. Polymer thickening is boosted via neutralization with bases (organic, inorganic, amino acids, etc.). Maximum viscosity is typically achieved at pH values of 6.0–7.0. Figure 2 illustrates the effect of pH on different concentrations of Carbopol 980 NF polymer dispersions.

Choice of Neutralizing Agent

A wide range of neutralizing agents can be used to achieve the target pH. Selection of a neutralizing agent is based upon the vehicle and intended characteristics of the finished product. Table 3 includes a list of the most commonly used neutralizers for aqueous systems. For hydroalcoholic systems, the choice of an effective neutralizer depends upon the level of alcohol in the system. Table 4 provides a list of recommended neutralizers for hydroalcoholic systems based upon the amount of alcohol present.
Additional information regarding the choice and use of a neutralizing agent can be found in the Bulletin: Thickening Properties and technical data sheet: Neutralizing Carbopol and Pemulen Polymers in Aqueous and Hydroalcoholic Systems at [www.lubrizol.com/health](http://www.lubrizol.com/health).

**Table 3. Commonly Used Neutralizers**

<table>
<thead>
<tr>
<th>Neutralizer/Amount</th>
<th>Neutralization Ratio (approximate amount per one part polymer) pH = 6–7</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sodium Hydroxide (18% solution)</td>
<td>2.30</td>
</tr>
<tr>
<td>Potassium Hydroxide (18% solution)</td>
<td>3.30</td>
</tr>
<tr>
<td>Ammonium Hydroxide (28% solution)</td>
<td>0.70</td>
</tr>
<tr>
<td>Triethanolamine</td>
<td>1.50</td>
</tr>
<tr>
<td>Tromethamine</td>
<td>1.30</td>
</tr>
<tr>
<td>Aminomethyl Propanol</td>
<td>0.95</td>
</tr>
</tbody>
</table>

**Table 4. Neutralizers for Hydroalcoholic Systems**

<table>
<thead>
<tr>
<th>Hydroalcoholic Systems (up to alcohol %)</th>
<th>Neutralizer</th>
</tr>
</thead>
<tbody>
<tr>
<td>20 – 30</td>
<td>Sodium Hydroxide or Potassium Hydroxide</td>
</tr>
<tr>
<td>60</td>
<td>Triethanolamine</td>
</tr>
<tr>
<td>80</td>
<td>Tromethamine</td>
</tr>
<tr>
<td>80</td>
<td>Aminomethyl Propanol</td>
</tr>
</tbody>
</table>

**Hydrogen Bonding**

For anhydrous systems or systems where pH adjustment is not feasible, formulation thickening can be enhanced through hydrogen bonding with hydroxyl donor co-excipients.

**Commonly Used Hydroxyl Donors:**

- Polyols (glycerin, propylene glycol and polyethylene glycol)
- Sugar alcohols (mannitol, sorbitol)
- Select nonionic surfactants
- Polyethylene oxide
Hydrogen Bonding (Continued)

A comparison of Figures 3 and 4 demonstrates the effects of hydrogen bonding on viscosity for varying concentrations of Carbopol polymers in anhydrous and aqueous media.

**Figure 3.** Viscosity of Carbopol Polymer Dispersions in Glycerin as Prepared - Hydrogen Bonding  

**Figure 4.** Viscosity of Carbopol Polymer Dispersions in Water as Prepared

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### Viscoelastic Behavior

#### Shear Thinning - Pour, Pump, Spread and Spray

The viscoelastic behavior of Lubrizol Carbopol polymers is ideal for pouring, pumping, spreading and spraying applications.

1. Viscosity is high when the system is at rest.
2. The application of shear forces decreases viscosity and facilitates flow.
3. After shearing stops, the high viscosity of the original system is quickly restored.

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### Suspension Stabilization

#### Yield Value vs. Viscosity

- **Yield Value** - The minimum amount of shear stress needed before flow begins.
- **Viscosity** - An expression of the resistance of a fluid to flow; the higher the viscosity, the greater the resistance.

While viscosity can only slow down the rate of settling, a high yield value is crucial to the successful formation of permanent suspensions.

The high yield values of Carbopol polymers enable the formation of permanent, stable suspensions at significantly low polymer concentrations. The polymers swell when hydrated and neutralized, forming a colloidal dispersion.

The swollen, closely packed microgels hold solid particles within the gel structure and result in the formation of a robust suspension.
Emulsion Stabilization – Pemulen Polymers

Pemulen polymers are carbomer copolymers of acrylic acid and a long-chain alkyl methacrylate crosslinked with allyl ethers of pentaerythritol. A combination of large hydrophilic and small lipophilic regions within the polymer matrix facilitates the formation of stable oil/water emulsions through steric and associative stabilization mechanisms.

- The hydrophilic portion of the polymer forms a gel network around the oil droplets.
- The hydrophobic portion functions as an anchor for the oil phase.

Pemulen Polymers vs. Traditional Surfactants

- Formulation flexibility: largely independent of oil type(s), hydrophilic-lipophilic balance (HLB) values and oil levels.
- Highly efficient: Typically used at concentrations of 0.20 – 0.40 wt% to achieve emulsion stabilization. If necessary, viscosity can be adjusted with Carbopol polymers.
- Use of a surfactant is not required; however, the polymer may be combined with low amounts of surfactant to achieve a smaller droplet size and whiter, brighter appearance.
- Increased viscosity and yield values with polymer use result in improved emulsion stability.
- Finished formulations benefit from shear thinning rheological properties.

The Skin Contact Advantage

Ions present in the skin trigger the release of oil droplets contained within the emulsion. The oil droplets coalesce to form a lipophilic film on the surface of the skin. In the absence of surfactants, this film cannot be re-emulsified or washed off. This property of Pemulen polymers is particularly advantageous when used in the formulation of sunscreen products.
Table 5. Specifications and typical properties of Pemulen polymers in product formulations.

<table>
<thead>
<tr>
<th>Pemulen™ Polymers</th>
<th>TR-1 NF</th>
<th>TR-2 NF</th>
</tr>
</thead>
<tbody>
<tr>
<td>USP/NF Monograph</td>
<td>Carbomer Copolymer Type B</td>
<td>Carbomer Copolymer Type A</td>
</tr>
<tr>
<td>Typical Oil Levels</td>
<td>Up to 20%-30%</td>
<td>Up to 50%</td>
</tr>
<tr>
<td>Typical Use Level (wt%)</td>
<td>0.20-0.40</td>
<td>0.20-0.40</td>
</tr>
<tr>
<td>pH Formulating Range</td>
<td>4-8</td>
<td>4-8</td>
</tr>
<tr>
<td>Viscosity (mPa.s) pH 7.3-7.8</td>
<td>6,500-15,500</td>
<td>1,700-4,500</td>
</tr>
<tr>
<td>0.2 wt% emulsion</td>
<td>14,000-26,500</td>
<td>4,500-13,500</td>
</tr>
</tbody>
</table>

Benefits of Carbopol ETD and Ultrez Polymers

- Processing versatility: Carbopol Ultrez 10 NF and Carbopol ETD 2020 NF polymers can be added directly to water without screening or the use of an eductor.
- Unique performance characteristics enable the polymers to wet quickly, yet hydrate slowly.
- Use minimizes particle agglomeration when turbulent mixing is not available for dispersion.
- Facilitate processing and handling: compared with traditional Carbopol polymers, Carbopol ETD and Ultrez polymers produce dispersions that are much lower in viscosity prior to neutralization.
- Upon neutralization, the polymers become highly efficient formulation thickeners.

Processing Considerations and Best Practices

Dispersion Preparation

The dispersion technique employed depends upon the size of the batch, the polymer concentration and the manufacturing equipment that is available.

Table 6. Dispersion Recommendations for Carbopol Polymers

<table>
<thead>
<tr>
<th>Approximate Batch Size</th>
<th>Traditional Carbopol® Polymer Concentration (%w/w)</th>
<th>Recommended Equipment</th>
</tr>
</thead>
<tbody>
<tr>
<td>Small-Medium</td>
<td>Up to ~2%</td>
<td>Coarse sieve or 20 mesh screen; a mixer with a conventional open-blade impeller</td>
</tr>
<tr>
<td>Several hundred milliliters to several hundred liters</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Large</td>
<td>Up to ~3%</td>
<td>Eductor or mechanical in-line disperser</td>
</tr>
<tr>
<td>Tens of hundreds of liters</td>
<td></td>
<td></td>
</tr>
<tr>
<td>Large - Continuous Production</td>
<td>Up to ~3%</td>
<td>Mechanical in-line powder disperser</td>
</tr>
<tr>
<td>Tens of hundreds of liters</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>
Processing Recommendations

- Use a stainless-steel screen, eductor or mechanical disperser to add the polymer to the aqueous phase to facilitate the formation of a uniform, stable dispersion.

- When possible, add the polymer at the beginning of the manufacturing process and neutralize at the end.

- Use water at room temperature to disperse traditional polymers.

- For best results, disperse the polymer in water that does not contain salts or alkalis. The presence or addition of electrolytes will significantly reduce the viscosity of the dispersion.

- After dispersion, allow time for complete polymer hydration while mixing at a slow rate.

- Uniformly disperse the polymer to avoid agglomeration. Dispersions that are not properly prepared may exhibit a grainy or nonuniform appearance, viscosity variation and/or formulation instability.

- Neutralizers should be gradually introduced to allow for pH equilibration. The gradual introduction of the neutralizer will optimize the viscosity consistency and stability of the formulation.

- Some ingredient incompatibilities can be circumvented through adjustments in the order of addition (e.g., electrolytes added after pH adjustment).

Emulsion Preparation

An indirect method of dispersion is recommended for emulsion preparation:

- Pre-disperse the polymer in the nonpolar (oil) phase before adding the aqueous phase.

- The polymers readily disperse in most oils.

- Once the oil dispersion is mixed with the aqueous phase, neutralize with a suitable base.

Active Pharmaceutical Ingredient (API) Addition

Incorporating an active pharmaceutical ingredient (API) into Carbopol polymer dispersions depends upon the physical/chemical properties of the API:

- Insoluble ingredients: can be added before or after the pH adjustment of the polymeric dispersion.

- Soluble ingredients: can be dissolved in the water used to prepare the polymeric dispersion. It may be beneficial to add some soluble ingredients to the final formulation in order to avoid compatibility issues (e.g., electrolytes are commonly added during the final phase of the dispersion preparation).
Mixing Guidelines

- Use a sufficient amount of time and mixing to ensure that the polymer is completely hydrated prior to adding other formulation components.

- Excessive/improper mixing during dispersion may result in air entrapment, viscosity variation and/or formulation instability. Air entrapment can be minimized by using a variable-drive motor. Once the Carbopol polymer has been dispersed, air entrapment can be minimized by repositioning the impeller and reducing the mixing speed. Allow the acid dispersion to stand to release entrapped air bubbles prior to neutralization.

- Moderate agitation is recommended.

- Any necessary high-intensity mixing should be completed prior to neutralization.

- Avoid high shear mixing with Waring blenders or rotor-stator homogenizers. Such mixing can damage the polymer and result in permanent loss of functionality.

Equipment Cleaning

- Manufacturing equipment should be promptly cleaned after processing carbomer dispersions.

- Gelled residue may be removed by powerwashing with warm water.

- If an excessive gel layer has formed, it may be collapsed using a dilute solution of salt (5% w/v).

- Any dry residue that remains on equipment after processing may be soaked for 10-30 minutes using warm (~65°C) dilute alkaline solutions and then removed with pressure washing.

- Recommended detergent solutions:
  - 2% solution of P3-cosa® CIP 95 (Ecolab GmbH & Co. OHG)
  - 0.2% solution of Extran® AP12 (EMD - Merck KGaA)

Storage and Handling Recommendations

Dry Powder Polymer

Chemical Stability

Carbopol polymer powders are chemically very stable under normal storage conditions. Studies indicate that the chemical properties of the material do not change significantly over a five-year period.

Moisture Effect

In terms of physical stability in a powder state, the moisture content (also referred to as loss on drying, or LOD) increases with storage time due to the hygroscopic nature of Carbopol polymers. A change in the moisture content of the polymer does not affect its chemical stability; however, it does affect the material’s ability to meet the LOD specification. The material should be periodically retested to ensure conformance with the loss on drying specification. Carbopol Polymers, Pemulen polymeric emulsifiers and Noveon AA-1 polycarbophil are hygroscopic products. They are supplied with moisture content of less than 2 percent as determined by loss on drying testing. Moisture pickup does not affect the efficiency of polymers, but polymers containing high levels of moisture are more difficult to disperse and weigh accurately. Therefore, containers of Carbopol and Pemulen polymers must be tightly closed and stored out of contact with water.

Thermal Stability

The polymers are thermally stable under normal conditions. When exposed to excessive temperatures above the glass transition temperature (~105°C), the dry polymers become sintered. Depending on the temperature and duration of exposure, the dry polymer may become discolored. Sintered polymers are more difficult to disperse and slower to gel in liquid formulations.
Microbial Growth

Carbopol polymers, Pemulen polymers and Noveon AA-1 polycarbophil do not support bacteria, mold or fungus growth in powder form.

Dust Precautions

Based on the physical characteristics of Lubrizol powders, routine safety precautions should be taken to prevent potentially explosive dusting.

Polymer Dispersions

Oxidation Stability

Lubrizol polymers are not subject to hydrolysis or oxidation under normal conditions. The viscosity of liquid formulations can decrease as a result of reactive oxygen. This reaction is catalyzed by UV light or trace levels of metals that may be present in the water. Chelating agents and UV absorbers can be used to minimize degradation.

Control of Microbial Growth

The polymers will not support mold or bacterial growth, but will not suppress it. For prolonged storage, use of an appropriate preservative is recommended.

Electrolytes and Cationics

Use of electrolytes or cationics as co-ingredients with Carbopol polymers may lower viscosity, reduce clarity and contribute to the precipitation of a formulation. The incorporation of multivalent vs. monovalent ions can increase the severity of these effects. Figure 5 demonstrates the typical effect of sodium chloride on Carbopol polymer viscosity. In general, lightly crosslinked Carbopol polymers perform better than highly crosslinked Carbopol polymers in electrolyte systems.

Recommendations to minimize cationic or electrolyte co-ingredient effects:

- Increase the Carbopol polymer concentration.
- Select a different grade of Carbopol polymer.
- Reduce the amount of electrolytes or cationics used in the formulation.
- Add electrolytes or cationics after pH is adjusted.
- Use cationics with a lower charge density and higher molecular weight.

Figure 5. Effect of Salt on Viscosity of 1.0% Carbopol Polymer Dispersions at pH 6.0
About Lubrizol Life Science Health
The Health business team partners with customers to speed their innovative medical devices and differentiated pharmaceutical products to market. Our dedicated team provides best-in-class polymers and excipients, along with state-of-the-art product design, development, and manufacturing services, with the ultimate goal of creating solutions that improve patient outcomes.

For more information, visit lubrizol.com/Health